# **Material Safety Data Sheet For Li-ion Batteries**

### Section 1— Product Identification

Product Name:Lithium Ion BatteryModel & Rating:/Chemical System:Li(NixCoyMn1-x-y)O2/CManufacturer Name:JYH Technology Co., Ltd.Manufacturer Name:No. 12, Bangmin Road, Jianghai District, Jiangmen City, Guangdong Province, P.R.ChinaPhone Number: +86-750-3808313Fax Number: +86-750-3808133

### Section 2—Composition /Information on Ingredients

Although the chemical composition of the various cell manufacturers is proprietary, the following is typical of the chemistry.

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Chemical Name	CAS No.	%
Aluminum Foil	7429-90-5	2-6
Copper Foil	7440-50-8	5-10
dimethyl carbonate (DMC)	616-38-6	5-17
Diethyl carbonate (DEC)	105-58-8	5-17
Ethylene Carbonate (EC)	96-49-1	5-17
Graphite Powder	7440-44-0	15-25
Lithium Nickel-Cobalt-Manganese Dioxide Li(Ni <sub>x</sub> Co <sub>y</sub> Mn <sub>1-x-y</sub> )O <sub>2</sub>	182442-95-1	25-40
Lithium Hexaflurorphosphate	21324-40-3	1-5
Poly Vinylidene Fluoride (PVDF)	24937-79-9	0.1-1
Nickel	7440-02-0	0.1-2
Iron	7439-89-6	5-15

### Section 3- Hazard Identification

### 3.1 Physical:

The Lithium Ion batteries described in this Material Safety Data Sheet are sealed which are not hazardous when used according to the recommendations of the manufacturer.

Under normal conditions of use, electrode materials and liquid electrolyte they contain are non-reactive provided the battery integrity is maintained and seals remain intact, Risk of exposure only in case of abuse, e.g. mechanical, thermal, electrical, which leads to the activation of safety valves and/or the rupture of the battery containers. Electrolyte leakage, electrode materials reaction with moisture/water of battery vent/ explosion/fire may follow depending upon circumstances.

### 3.2 Chemical:

#### Classification of dangerous Substances Contained into the Product as per Directive

	-				-		
Substance	Chemical	Melting Point	Boiling	Exposure	Indication	Special	Safety
	Symbol		point	limit	of Danger	Risk*	Advice**
Lithium	Li(Ni <sub>x</sub> Co <sub>y</sub>	>500°C		0.1 mg/m3		R22R43	S2 S22 S24
Nickel-	Mn1-x-y)O2			OSHA			S26
Cobalt-							S36 S37 S45
Manganese							
Dioxide							
Organic	EC	EC: 38°C	EC: 243°C	None	Flammable	R21R22	S2 S24
Solvents	DMC	DMC: 4°C	DMC: 90°C	established		R41	S26 S36
	DEC	DEC: -43°	DEC :	OSHA		R42/43	S37 S45
			127°C				

## **Material Safety Data Sheet For Li-ion Batteries**

	LiPF <sub>6</sub>	N/A	N/A	None	Irritant	R14R21	S2 S8 S22
		(decomposes		established	Corrosive	R22	S24 S26 S36
		at 160°C)		OSHA		R41	S37 S45
						R43	

### \*: Name of Special Risks:

R14/15 Reacts with water and yields flammable gases

R21 Harmful in contact with skin

R22 Harmful us swallowed

R35 Causes severe burns

R41 Risk of serious damage to the eye

R42/43 May cause sensitization by inhalation and skin contact

R43 May cause sensitization by skin contact

### \*\*: Name of Safety Advices:

S2 Keep out of reach from children

S8 Keep away from moisture

S22 Do not breathe dust

S24 Avoid contact with skin

S26 In case of contact with eyes, rinse immediately with plenty of water and seek medical attention

S36 Wear suitable protective clothing

S37 Wear suitable gloves

S45 In case of incident, seek medical attention

### Section 4- First Aid Measures

In case of battery rupture or explosion, evacuate personnel from contaminated area and provide maximum ventilation to clear out corrosive fumes/gases and pungent odors.

In all case, seek immediate medical attention,

Eye contact: Flush with plenty of water (eyelids-held open) for at least 15 minutes

Skin contact: Remove all contaminated clothing and flush affected areas with plenty of water and sop for at least 15minutes.

Ingestion: Dilute by giving plenty of water and get immediate medical attention.

Assure that the victim does not aspirate vomited material by use of positional drainage.

Assure that mucus does not obstruct the airway.

Do not give anything by mouth to an unconscious person

Inhalation: Remove to fresh air and ventilate the contaminated area.

Give oxygen or artificial respiration if needed.

### Section 5-Fire and explosion Hazard Data

ĺ	Fire and explosion hazard	The batteries can leak and/or spout vaporized or decomposed and
		combustible electrolyte fumes in case of exposure above 90°C resulting
		from inappropriate use or from the environment. Possible formation of
		hydrogen fluoride (HF) and phosphorous oxides during fire.LiPF6 salt
		contained in the electrolyte releases hydrogen fluoride (HF) in contact with
		water.

# **Material Safety Data Sheet For Li-ion Batteries**

Extinguishing media	Suitable: CO <sub>2</sub> ,
	Dry chemical or Foam extinguishers
	Not to be used : Type D extinguishers
Special exposure hazards:	Following cell overheating due to external source or due to improper use,
	electrolyte leakage or battery container rupture may occur and release inner
	component/material in the environment.
	Eye contact: The electrolyte solution contained in the battery is irritant to
	ocular tissues.
	Skin contact: The electrolyte solution contained in the battery causes skin
	irritation.
	Ingestion: The ingestion of electrolyte solution causes tissue damage to
	throat and gastro/respiratory tract.
	Inhalation: Contents of a leaking or ruptured battery can cause respiratory
	tract, mucus, membrane irritation and edema.
Special protective	Use self-contained breathing apparatus to avoid breathing irritant fumes.
Equipment	Wear protective clothing and equipment to prevent body contact with
	electrolyte solution.

### Section 6- Accidental Release or Spillage

Personal Precautions,	Restrict access to area until completion of clean-up. Do not touch the spilled
protective equipment, and	material. Wear adequate personal protective equipment as indicated in
emergency procedures	Section 8.
Environmental Precautions	Prevent material from contaminating soil and from entering sewers or
	waterways.
Methods and materials for	Stop the leak if safe to do so. Contain the spilled liquid with dry sand or
Containment	earth. Clean up spills immediately.
Methods and materials for	Absorb spilled material with an inert absorbent (dry sand or earth). Scoop
cleaning up	contaminated absorbent into an acceptable waste container. Collect all
	contaminated absorbent and dispose of according to directions in Section
	13.
	Scrub the area with detergent and water; collect all contaminated wash
	water for proper disposal.

### Section 7-Handling and Storage

The batteries should not be opened destroyed nor incinerated since they may leak or rupture and release in the environment the ingredients they contain.

Handling	Do not crush, pierce, short (+) and (-) battery terminals with conductive (i.e. metal) goods.
	Do not directly heat or solder. Do not throw into fire.
	Do not mix batteries of different types and models. Do not mix new and used batteries.
	Keep batteries in non-conductive (i.e. plastic) trays.
Storage	Store in a cool (preferably below 30°C) and ventilated area away from moisture, sources of
	heat, open flames, food and drink.
	Keep adequate clearance between walls and batteries.

# **Material Safety Data Sheet For Li-ion Batteries**

	Temperature above 90°C may result in battery leakage and rupture. Since short circuit can	
	cause burn, leakage and rupture hazard, keep batteries in original packaging until use and	
	do not jumble them.	
Other	Manufacturer recommendations regarding maximum recommended currents and operating	
	temperature range.	
	Applying pressure on deforming the battery may lead to disassembly followed by eye, skin	
	and throat irritation.	

### Section 8-Exposure Controls / Person Protection

Respiratory protection	Not necessary under normal use.	
	In case of battery rupture, use self-contained full-face respiratory equipment.	
	equipment with type ABEK filter.	
Hand protection	Not necessary under normal use.	
	Use rubber gloves if handling a leaking or ruptured battery.	
Eye protection	Not necessary under normal use. Wear safety goggles or glasses with	
	side shields if handling a leaking or ruptured battery.	
Skin protection	Not necessary under normal use. Use rubber apron and protective	
	working in case of handling of a ruptured battery.	

### Section 9—Physical and Chemical Data

9.1 Appearance (Physical shape and color as supplied:)

Li(Ni <sub>x</sub> Co <sub>y</sub> Mn <sub>1-x-y</sub> )O <sub>2</sub>	Gray odorless power
Graphite	Black or odorless power
Organic solvent	Colorless liquid
Lithium salt	White, crystalline and odorless power
9.2 Specific gravity (H <sub>2</sub> O=1)	
Li(Ni <sub>x</sub> Co <sub>y</sub> Mn <sub>1-x-y</sub> )O <sub>2</sub>	3.6-3.9
Graphite	2.0-2.2
9.3 Melting point	
Li(NixCoyMn1-x-y)O2	1130°C
Graphite	3500-3900°C
9.4 Flash point	Not available

#### Section 10- Stability and Reactivity Data

Stability	Stable under normal temperatures and pressures.
Conditions to avoid	Heat above 70°C or incinerate. Deform, mutilate, crush, pierce, disassemble.
	Short circuit. Prolonged exposure to humid conditions.
Materials to avoid	N/A
Hazardous	Corrosive/Irritant Hydrogen fluoride (HF) is produced in case of reaction of
decomposition	lithium (LiPF <sub>6</sub> ) with water.
products	Combustible vapors and formation of Hydrogen fluoride (HF) and phosphorous
	oxides during fire.
	Toxic fumes, and may form peroxides.

## **Material Safety Data Sheet For Li-ion Batteries**

#### Section 11- Toxicological information

Irritation	In the event of exposure to internal contents, vapor fumes may be very irritating	
	to the eyes and skin.	
Sensitization	No data is available	
Reproductive toxicity	No data is available	
Toxicologically	No data is available	
synergistic material		

### Section 12-Ecological Information

General note	Do not allow undiluted product or large quantities of it to reach ground water,		
	water course or sewage system.		
Mobility in soil	No data is available		
Persistence and	No data is available		
degradability			

### Section 13-Disposal Method

Dispose in accordance with applicable regulations which vary from country to country.

(In more countries, the thrashing of used batteries is forbidden and the end-users are invited to dispose them properly, eventually through not-for-profit organizations, mandated by local governments or organized on a voluntary basis by professionals).

Lithium Ion batteries should have their terminals insulated and be preferably wrapped in plastic bags prior to disposal.

- 13.1 Incineration: Incineration should never be performed by battery users but eventually by trained professionals in authorized facilities with proper gas and fumes treatment.
- 13.2 Land filling: Leach ability regulations (mg/l)

[	Component	Leach ability	EC limit	EPA	Other*
	Iron	100	2		5
	Nickel	500			0.5

13.3 Recycling: Send to authorized recycling facilities, eventually through licensed waste carrier.

### Section 14-Transportation Information

14.1 UN number of Lithium Ion batteries: UN3480 or UN3481.

- 14.2 Lithium Ion batteries have been tested under provisions of the UN Manual of Tests and Criteria, the batteries are passed the UN 38.3 test, Part III, sub-section 38.3 (withstanding a 1.2m drop test) and are classified as non-dangerous goods.
- 14.3 The package of battery by air should be complied with the requirements of Packing Instruction 965 (Section II/ Section IB) or 966/967 (Section II) of IATA DGR 63th Edition for transportation. The battery containing Watt-hour rating by air is not more than 100Wh.
- 14.4 The battery is not restricted according to IMO IMDG Code (inc Amdt 40-20). Need to meet the Special Provision International Maritime dangerous goods code (IMDG) 188, 230, 310, 348, 360, 376, 377.
- 14.5 The packaging is required to be strong and can prevent the products from short-circuit.

With regard to transport, the following regulations are cited and considered:

#### 5 / 6

## **Material Safety Data Sheet For Li-ion Batteries**

- The International Civil Aviation Organization (ICAO) Technical Instructions.
- The International Air Transport Association (IATA) Dangerous Goods Regulations.
- The International Maritime Dangerous Goods (IMDG) Code.
- The US Hazardous Materials Regulation (HMR) pursuant to a final rule issued by RSPA
- The Office of Hazardous Materials Safety within the US Department of Transportations' (DOT) Research and Special Programs Administration (RSPA).

### Section 15-Regulatory Information

The transport of rechargeable Lithium Ion batteries is regulated by various bodies (IATA, IMO, ADR, US-DOT) that follow the United Nations "Recommendations on the Transport of Dangerous Goods, Model Regulations, 21th Revised edition - Ref.ST/SG/AC.10/1 Rev. 21".

Depending on their lithium metal equivalent weight content, design, and ability to pass safety tests defined by the UN in the "Recommendations on the Transport of Dangerous Good - Manual of Tests and Criteria – 7<sup>th</sup> Revised edition - Ref. ST/SG/AC.10/11 Rev.7 Amendment 1 «Lithium Batteries»", the Lithium Ion cells and the battery packs are not be assigned to the UN N°3480/3481 Class-9, that is restricted for transport.

### Section 16-Other Information

This information has been compiled from sources considered to be dependable and is, to the best of our knowledge and belief, accurate and reliable as of the date compiled. However, no representation, warranty (either expressed or implied) or guarantee is made to the accuracy, reliability or completeness of the information contained herein. This information relates to the specific materials designated and may not be valid for such material used in combination with any other materials or in any process. It is the user's responsibility to satisfy himself as to the suitability and completeness of this information for his particular use.

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